

**Charge & Electronegativity – Guided Notes**

Electricity is a \_\_\_\_\_.

Static Electricity is

\_\_\_\_\_  
\_\_\_\_\_.

The protons and neutrons are \_\_\_\_\_ in the nucleus and \_\_\_\_\_.

Electrons are bound \_\_\_\_\_. These are the particles that can “\_\_\_\_\_” an object.

**Conservation of Charge says:**

**What is electronegativity?**

When you rub two objects with very different electronegativities together, you will \_\_\_\_\_

\_\_\_\_\_.

**Example Electronegativity Series:**

Glass (electron donors)

\_\_\_\_\_

\_\_\_\_\_

\_\_\_\_\_

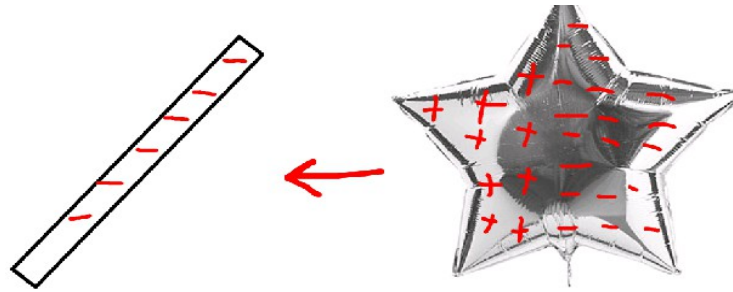
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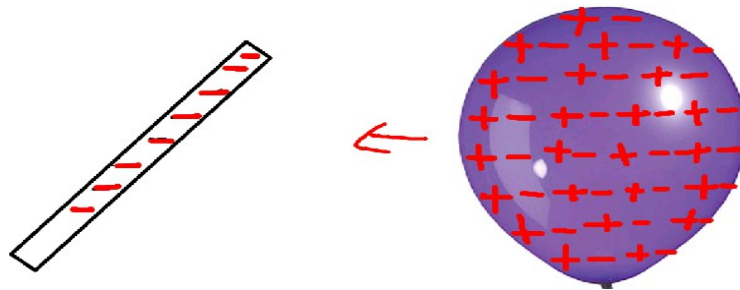
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PVC (electron acceptors)

Why does the foil attract?



Why does the latex balloon attract?



Three Types of Charge Transfer:

1. Friction: \_\_\_\_\_

2. Conduction: \_\_\_\_\_

3. Induction: \_\_\_\_\_

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